CHM 22	00C	
February	y 23,	2005

Quiz 6

Name ___KEY (Please Print)

Grade __10___ of 10 points

Points

- 1. (a). Give both the **common name** and the **IUPAC** name for the following ether.
- (2) 1 pt each. -0.5 pts if groups not in alphabetical order.

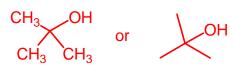
Common Name <u>t-butyl ethyl ether</u>

IUPAC Name 2-ethoxy-2-methylpropane

(b) This ether can be synthesized by the Williamson reaction according to the following (2) scheme: 1 pt each structure

ROH + Na
$$\rightarrow$$
 RO $^{-}$ Na $^{+}$ + $^{1}/_{2}$ H₂
RO $^{-}$ Na $^{+}$ + R'Br \rightarrow ROR' + NaBr

where **ROR'** represents the ether. Draw the structure of **ROH** and **R'Br**:



ROH

R'Br

(3) 2. Fill in the blanks in the following statement with the letter corresponding to the items in the key list at the right.

0.5 pts each blank

A. aldehydes

The **Tollens test** is a test for __A__(A,B,C, or D?) which are

 $\underline{\underline{K}}$ (K or L?) to $\underline{\underline{D}}$ (A,B,C, or D?) in the test.

The reagent used in the test is _____(E,F,G,H,I, or J?) which is

<u>L</u> (K or L?) to <u>H</u> (E,F,G,H,I, or J?)

in the reaction.

- B. ketones
- C. alcohols
- D. carboxylic acids
- E. $Cr_2O_7^2$
- F. Cr³⁺
- G. Ag⁺
- H. Ag(s)
- I. Cu²⁺
- J. $Cu_2O(s)$
- K. oxidized
- L. reduced
- (3) 3. Indicate the products of the following acid-base reactions. If the reaction does not proceed predominantly to the right, enter No Reaction as the product.

1 pt each

$$CH_3COOH$$
 (acetic acid) + $OH^- \rightarrow CH_3COO^- + H_2O^-$

$$CH_3CH_2OH (ethanol) + OH^- \rightarrow No Reaction$$

$$C_6H_6OH (phenol) + OH^- \rightarrow C_6H_6O^- + H_2O^-$$